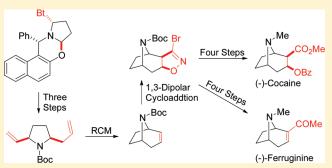
Total Synthesis of (–)-Cocaine and (–)-Ferruginine

Guolin Cheng, Xinyan Wang,* Rui Zhu, Changwei Shao, Jimin Xu, and Yuefei Hu*

Department of Chemistry, Tsinghua University, Beijing 100084, People's Republic of China

Supporting Information

ABSTRACT: Total synthesis of tropane alkaloids (-)-cocaine and (-)-ferruginine were accomplished in nine steps each and in 55% and 46% overall yields, respectively, starting from the known Betti base derivative (+)-(7aR,10R,12S)-10-(1H-benzotriazol-1-yl)-7a,8,9,10-tetrahydro-12-phenyl-12H-naphtho-[1,2-e]pyrrolo[2,1-b][1,3]oxazine. In this novel route, RCM reaction and 1,3-dipolar cycloaddition were employed as key steps for the enantioselective construction of tropane skeleton and the regioselective introduction of 3-bromo-2-isoxazoline ring as masked *cis*-2,3-disubstituents. To obtain the desired precursor (2*S*,5*R*)-2-allyl-5-vinylpyrrolidine for RCM reaction, we developed a general and practical method for the prepara-



tion of enantiopure *cis*-2,5-disubstituted pyrrolidines bearing alkene- and/or alkyne-containing substituents. We also offered two highly efficient pathways for the conversion of the 3-bromo-2-isoxazoline ring into the desired *cis*-2,3-disubstituted groups in (-)-cocaine and (-)-ferruginine.

INTRODUCTION

Many tropane alkaloids exhibit biologically important properties and serve as lead compounds in drug discovery. As shown in Chart 1, the natural (-)-cocaine $(1)^{1}$ is a serotonin-norepinephrine-dopamine reuptake inhibitor, which mediates functionality of these neurotransmitters and leads to cocaine abuse. The unnatural (-)-ferruginine $(2)^{2}$ has proved to be a potent agonist for nicotinic acetylcholine receptor (nAchR), although its natural enantiomer has a negligible affinity for nAchR. For a long time, their typical structures and biologically important properties have made them highly attractive targets for total synthesis.³

Both alkaloids 1 and 2 have identical nonracemic tropane skeleton, but different substituents on C2 and C3. In fact, the first asymmetric synthesis⁴ of 2 was started from 1 in 1977. In recent years, many asymmetric routes have been reported for total synthesis of 1, 5, 2, 6 and their enantiomers by using different strategies. On the basis of the published works, a highly efficient route may be expected to have three key points: (a) highly stereoselective construction of the tropane skeleton; (b) highly regioselective introduction of cis-2,3-disubstituents on the tropane ring; and (c) highly efficient functional group transformations. According to this opinion, two references attracted our attention as shown in Scheme 1. Aggarwal^{6b} reported that the tropene derivative 4 could be stereoselectively constructed by RCM reaction of cis-2-allyl-5-vinylpyrrolidine derivative 3b. Rapoport^{5d} reported that 1,3-dipolar cycloaddition of 4 yielded the tricyclic 5a, in which the cis-2,3-disubstituents on the tropane ring were regioselectively introduced as an 2-isoxazoline ring. However, these two excellent transformations have never been used together for total synthesis of 1 or 2, most possibly because the preparation of 3b is still a challenging project to date and the functional group transformations of 5a are associated with tedious performances.

Herein, we would like to report a novel route for the total synthesis of 1 and 2, in which RCM reaction and 1,3-dipolar cycloaddition mentioned in Scheme 1 were used together as key steps. Meanwhile, we also developed a practical method for the preparation of enantiopure 3b and offered the 2-isoxazoline derivative 5b as a versatile precursor, by which 1 and 2 were synthesized by controlling selectivity with highly efficient functional group transformations.

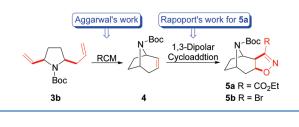
RESULTS AND DISCUSSION

Efficient Preparation of Enantiopure (2*S*,*SR*)-2-Allyl-5vinyl-*N*-Boc-pyrrolidine (3b). In recent years, RCM reaction of *cis*-2,5-disubstituted pyrrolidines has emerged as an efficient method in the synthesis of azabicyclics.^{6b,7,8} Depending on whether the substituent was alkene or alkyne, as well as the chain lengths, different sized and substituted azabicyclics were prepared. However, the method was rarely employed to construct the nonracemic azabicyclics because the preparation of the corresponding nonracemic *cis*-2,5-disubstituted pyrrolidines still remains a challenging task to date.^{6b,8,9} In a few examples,^{6b,8} the RCM reaction was so efficient that the method was indeed the preparation of the corresponding nonracemic *cis*-2,5-disubstituted pyrrolidines.

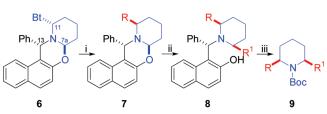
```
Received:January 11, 2011Published:March 10, 2011
```



Scheme 1







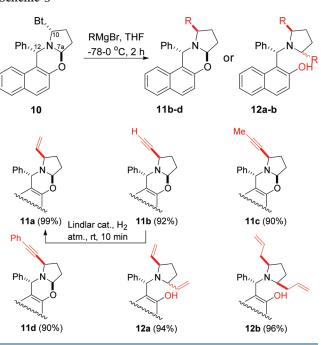
Bt = Benzotriazol-1-yl; R, R¹ = alkene- or alkyne-containing substitutents

 a Conditions: (i) RMgBr, THF, 0 °C; (ii) RMgBr, Et_2O, 0 °C; (iii) (a) aq NaOH, MeOH, THF, 60 °C, (b) Boc_2O, K_2CO_3, CH_2Cl_2, rt.

Recently, we reported a general three-step method¹⁰ for the preparation of enantiopure *cis*-2,6-disubstituted piperidines bearing alkene- and/or alkyne-containing substituents from Betti base derivative **6**.¹¹ As shown in Scheme 2, when **6** underwent two solvent-controlled regioselective alkylations, the alkene- or alkyne-containing substituent was introduced to C11 (7) or C7a (**8**), respectively. Finally, the desired product **9** was obtained by a novel base-catalyzed *N*-debenzylation of **8**. Encouraged by this success, we tried to use the same strategy to prepare the required (2*S*,*SR*)-2-allyl-5-vinyl-*N*-Boc-pyrrolidine (**3b**).

Thus, Betti base derivative 10, which contains a pyrrolidine ring, was prepared in 93% yield as a single diastereomer by the reported procedure.11d Unfortunately, no solvent-controlled regioselectivity was observed in the alkylation of 10 with vinyl or allyl Grignard reagent. As shown in Scheme 3, when 10 was treated with H₂C=CHMgBr in THF below -40 °C, no reaction occurred at all. Instead of the expected 11a, a mixture of trans- and cis-dialkylated product 12a was obtained when the reaction proceeded above -40 °C. Although a single product was obtained in the alkylation of 10 with H_2C =CHCH₂MgBr at -78 °C, it was assigned as dialkylated product 12b. These results may be caused by the fact that the pyrrolidine ring in 10 has a planar configuration, which led to higher reactivity and less selectivity. Thus, several less active organometallic reagents were tested for the alkylation of 10, such as (H₂C=CH)₂CuLi (from H₂C=CHLi and CuI in situ), $(H_2C=CH)_2Zn$ (from $H_2C=CHMgBr$ and $ZnBr_2$ in situ), and H₂C=CHCH₂TMS/BF₃·Et₂O. However, the former one was inert at 0 °C or gave a complicated mixture along with unreacted





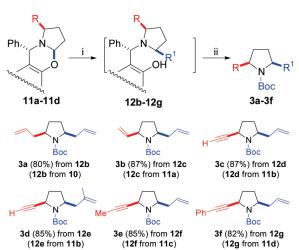
ARTICLE

10 at 25 °C. The later two could stop at the monoalkylation stage regioselectively at -40 °C, but the epimeric mixtures were obtained [3:1 ratio of **11a** and its epimer for $(H_2C=CH)_2Z_1$; 3:2 ratio of (R)- and (S)-allyl substituted on C10 for $H_2C=CHCH_2TMS/$ $BF_3 \cdot Et_2O$]. On the basis of these results, we suggested that the alkylation may go through an iminium salt pathway that may be caused by the Lewis acid properties of the ZnBr₂ and BF₃·Et₂O. Therefore, Nu⁻ could attack both sides of the iminium salt intermediate to give an epimeric mixture. Luckily, when 10 was treated with HC \equiv CMgBr in THF at -40 °C, a highly regioselective and diastereoselective alkylation occurred on C10 to give 11b in 92% yield as a single product. Similarly, 11c and 11d were obtained when MeC=CMgBr and PhC=CMgBr were used as alkylation reagents, respectively. These alkylations may go through the S_N2 mechanism because the Bt-group in 10 was replaced by an alkynyl group with complete inversion of configuration. Finally, the required intermediate 11a was obtained in almost quantitative yield by hydrogenation of 11b over Lindlar catalyst.

As shown in Scheme 4, when 11a-d were treated with different allyl Grignard reagents in Et₂O, 2,5-disubstituted pyrrolidine intermediates 12c-g were obtained smoothly. Since 12b-g showed ambiguous NMR spectra caused by the unusual coalescence phenomenon,^{11c} they were directly used in the next step without characterization. As was expected, their *N*-debenzylations proceeded much easier than that of the counterpart 8. When the mixtures of 12b-g in aq NaOH–MeOH–THF (1:2:2 by v/v)¹⁰ were heated at 60 °C for 1 h, the *N*-debenzylations of 12b-g were finished completely. The crude products were then captured by Boc₂O in situ to give 2,5-disubstituted *N*-Boc-pyrrolidines 3a-f in high yields for two steps. Since the two chiral carbons were formed in separated steps, 3a-f were easily confirmed to be enantiopure products by their ¹H and ¹³C NMR spectra.

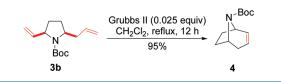
Efficient Preparation of the Tropene Derivative 4 by RCM and the Versatile Precursor 3-Bromo-2-isoxazoline Derivative 5b by 1,3-Dipolar Cycloaddition. In practice, the precursor





^{*a*} Conditions: (i) $H_2C=CHCH_2MgBr$ or $H_2C=C(Me)CH_2MgBr$, Et₂O, -40 °C, 1 h; (ii) (a) 6 M aq NaOH, MeOH, THF, 60 °C, 1 h, (b) Boc₂O, K₂CO₃, CH₂Cl₂, rt, 30 min, 80–87% yields for two steps.

Scheme 5



3b can be prepared in gram scale within a few hours from the intermediate **10**. As shown in Scheme 5, when **3b** was treated with Grubbs II catalyst in refluxed CH_2Cl_2 for 12 h, the desired tropene derivative **4** was obtained in 95% yield.

In 1998, Rapoport^{Sd} reported that 1,3-dipolar cycloaddition between 4 and $EtO_2C(CI)C$ =NOH yielded 3-carboethoxy-2isoxazoline derivative 5a. Since the 2-isoxazoline ring can be opened easily, it in fact is a masked *cis*-2,3-disubstituent. However, 5a seriously suffered from tedious functional group transformations to remove an extra carbon under very mild conditions, by which the epimerization of the 3-carbomethoxy group [which occupies the thermodyamically unfavorable axial position in (–)-cocaine (1)] could be avoided. Since this drawback comes from the structural nature of the reagent EtO₂C-(CI)C=NOH rather than from the strategy, we believed that it may be overcome by using an alternative reagent.

In the further investigation, the one-carbon 1,3-dipolar precursor dibromoformaldoxime (Br₂C=NOH) attracted our attention, which can take place by 1,3-dipolar cycloaddition with alkenes to yield the corresponding 3-bromo-2-isoxazolines under base conditions.^{12–14} The most important to us is that 3-bromo-2-isoxazoline can be converted into the β -hydroxy ester¹³ or β -hydroxy nitrile¹⁴ by different ring-opening methods. By a series of conditional experiments, we found that the regioselectivity of 1,3-dipolar cycloaddition between 4 and Br₂C=NOH significantly benefited from the reaction temperature. When it proceeded at 10 °C, the desired precursor **5b** was prepared in 91% yield (Scheme 6). As shown in Figure 1, the structure and stereochemistry

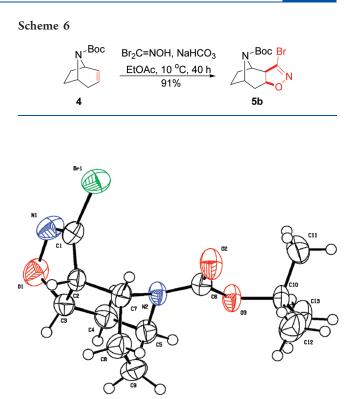


Figure 1. The structure and stereochemistry of 5b.

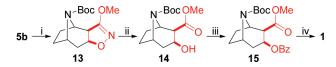
of **5b** was further confirmed by the single-crystal X-ray diffraction analysis.

Total Synthesis of (–)-Cocaine. It has been reported that 3-bromo-2-isoxazolines can be easily converted into 3-alkoxy-2isoxazolines under base conditions.^{12a,c,d,14c} As shown in Scheme 7, when the mixture of **5b** and NaOMe in MeOH was refluxed for 8 h, 3-methoxy-2-isoxazoline derivative **13** was obtained in 99% yield. In literature, the reductive cleavage of 3-methoxyisoxazoline into the corresponding β-hydroxy ester can be affected by several protocols.¹³ We found that the Raney-Ni-catalyzed hydrogenation originally reported by Curran^{13d,15} was the most practical one, by which the quantitative conversion of **13** into **14** was achieved in the presence of H₃BO₃. After **14** was esterified with BzCl, its product **15** was subjected to carry out a deprotection and a reductive amination in one-pot to give (–)-cocaine (**1**) in excellent yield. Thus, the total synthesis of (–)-cocaine (**1**) was accomplished in nine steps and in 55% overall yield (starting from **10**).

Total Synthesis of (–)-**Ferruginine (2).** To obtain the best result, three practical protocols for the conversion of 3-bromo-2-isoxazoline into β-hydroxyl nitrile were tested. When **5b** was treated with NaSEt^{14d} or NaI-TMSCl,^{14b} the former gave the desired product **16** in moderate yield (63%) and the latter led to a mixture because *N*-Boc was partially cleaved. However, when **5b** was hydrogenated in the presence of Raney-Ni and H₃BO₃,^{14c,15} **16** was obtained in quantitative yield (Scheme 8).

Although a reported conversion of β -hydroxyl nitrile into α,β -unsaturated nitrile with excess MeMgCl is very attractive,¹⁶ it failed in our hands. Thus, 17 was prepared in 85% yield by treatment of 16 with MsCl in the presence of Et₃N. Under the mild conditions, α,β -unsaturated ketone 18 was obtained in 87% yield by reaction of 17 with an excess of MeLi followed by a PPTS-catalyzed hydrolysis. In the one-pot reaction, 18 was

Scheme 7^{*a*}



^a Conditions: (i) NaOMe, MeOH, reflux, 8 h, 99%; (ii) Raney-Ni, H₂ (balloon), H₃BO₃, aq MeOH, rt, 3 h, 100%; (iii) BzCl, DMAP, Et₃N, CH₂Cl₂, rt, 12 h, 96%; (iv) (a) TFA, CH₂Cl₂, rt, 1 h, (b) 37% aq CH₂O, NaBH₃CN, rt, 1 h, 85%.





^{*a*} Conditions: (i) Raney-Ni, H₂ (balloon), H₃BO₃, aq MeOH, rt, 3 h, 99%; (ii) MsCl, Et₃N, CH₂Cl₂, 0 °C, 1 h, then rt, 12 h, 85%; (iii) (a) MeLi, Et₂O, 0–25 °C, 2 h, (b) aq PPTS, rt, 12 h, 87%; (iv) (a) TFA, CH₂Cl₂, rt, 1 h, (b) 37% aq CH₂O, NaBH₃CN, rt, 1 h, 91%.

subjected to carry out a deprotection and a reductive amination to give (-)-ferruginine (2) in excellent yield. Thus, the total synthesis of (-)-ferruginine (2) was accomplished in nine steps and in 46% overall yield (starting from 10).

CONCLUSION

A general and practical method for the preparation of enantiopure *cis*-2,5-disubstituted pyrrolidines bearing alkene- and/ or alkyne-containing substituents (3a-f) was developed. The desired tropene derivative 4 was constructed in excellent yield by RCM reaction of (2S,SR)-2-allyl-5-vinylpyrrolidine (3b). When 4 reacted with dibromoformaldoxime, the 1,3-dipolar cycloadduct 5b was obtained regioselectively. The 3-bromo-2isoxazoline unit in 5b has been proved to be an excellent masked *cis*- β -hydroxy ester or *cis*- β -hydroxyl nitrile. Finally, total synthesis of the tropane alkaloids (-)-cocaine (1) and (-)-ferruginine (2) was accomplished in nine steps each from 10. The novel route was characterized to employ the RCM reaction and 1,3-dipolar cycloaddition together as key steps and had the highest overall yields.

EXPERIMENTAL SECTION

Preparation of (7aR, 10R, 12S)-7a, 8, 9, 10-Tetrahydro-10-vinyl-12-phenyl-12H-naphtho[1,2-e]pyrrolo[2,1-b][1,3]oxazine (11a). A stirred mixture of 11b (6.50 g, 20 mmol) and 5% Lindler catalyst (16.3 mg, 2.5 wt %) in THF (50 mL) was hydrogenated at room temperature and atmospheric pressure until the hydrogen-absorbing rate was suddenly reduced (ca. 10 min). After the catalyst was filtered out, the filtrate was evaporated to give product 11a (6.48 g, 99%) as colorless crystals. Mp 103–105 °C (Et₂O); $[\alpha]_{D}^{25}$ +231.4 (c 0.2, CHCl₃); IR ν 3059, 2976, 1620, 1597 cm⁻¹; ¹H NMR δ 7.70–7.63 (m, 2H), 7.33-7.26 (m, 1H), 7.22-7.12 (m, 7H), 7.08 (d, J = 8.7 Hz, 1H), 5.66-5.55 (m, 1H), 5.42 (s, 1H), 5.19-5.07 (m, 2H), 4.88-4.83 (m, 1H), 3.86–3.79 (m, 1H), 2.14–1.92 (m, 3H), 1.90–1.78 (m, 1H); ^{13}C NMR δ 153.0, 142.6, 141.1, 131.5, 128.9 (2C), 128.8 (2C), 128.3, 128.2 (2C), 127.0, 126.2, 123.0, 122.9, 119.1, 116.9, 114.2, 85.6, 64.6, 54.8, 30.4, 28.7; MS m/z (%) 327 (M⁺, 6.49), 231 (100). Anal. Calcd for C23H21NO: C, 84.37; H, 6.46; N, 4.28. Found: C, 84.60; H, 6.39; N, 4.16.

ARTICLE

Preparation of (7aR, 10R, 12S)-7a, 8, 9, 10-Tetrahydro-10ethynyl-12-phenyl-12H-naphtho[1,2-e]pyrrolo[2,1-b][1,3]**oxazine (11b).** To a solution of **10** (2.09 g, 5 mmol) in dry THF (50 mL) was added a solution of HC=CMgCl in THF (1.0 M, 25 mmol) dropwise under nitrogen at -40 °C. After the reaction was stirred at 0 °C for 1 h (monitored by TLC), a saturated aqueous solution of NH_4Cl (10 mL) was added to quench the reaction. The resulting mixture was then extracted with Et₂O (2 \times 30 mL). Combined organic layers were washed with brine (2 \times 30 mL) and dried over anhydrous Na₂SO₄. After the removal of the solvent, the residue was purified by recrystallization to give product **11b** (1.49 g, 92%) as colorless crystals. Mp 160–162 °C (Et₂O); $[\alpha]_{D}^{25}$ +294.5 (c 0.2, CHCl₃); IR ν 3205, 2935, 2104, 1622, 1599 cm⁻¹; ¹H NMR δ 7.71–7.66 (m, 2H), 7.40-7.32 (m, 1H), 7.26-7.16 (m, 7H), 7.09 (d, J = 8.9 Hz, 1H),5.65 (s, 1H), 5.04 (d, J = 4.8 Hz, 1H), 4.20 - 4.15 (m, 1H), 2.30 - 2.14 (m, 3H), 2.11–1.96 (m, 1H), 1.84 (d, J = 2.0 Hz, 1H); ¹³C NMR δ 152.9, 142.4, 131.7, 129.0, 128.9, 128.8, 128.7 (2C), 128.4 (2C), 127.2, 126.2, 123.1, 122.9, 118.8, 113.1, 84.9, 84.2, 71.6, 55.7, 51.1, 30.6, 29.4; MS $m/z \ (\%)$ 325 (M⁺, 4.88), 231 (100). Anal. Calcd for C₂₃H₁₉NO: C, 84.89; H, 5.89; N, 4.30. Found: C, 84.62; H, 6.01; N, 4.42.

By a similar procedure as that used for 11b, compounds 11c,d were prepared.

Preparation of (7*aR*,10*R*,12*S*)-7*a*,8,9,10-Tetrahydro-10-(prop-1-ynyl)-12-phenyl-12*H*-naphtho[1,2-*e*]pyrrolo[2,1-*b*]-[1,3]oxazine (11c). By reaction of 10 with MeC≡CMgCl at -15 °C for 2 h, 11c was obtained in 90% yield. Mp 144–146 °C (Et₂O); [α]²⁵_D +378.9 (*c* 0.2, CHCl₃); IR ν 2993, 2962, 2195, 1622, 1597 cm⁻¹; ¹H NMR δ 7.75–7.65 (m, 2H), 7.43–7.36 (m, 1H), 7.28–7.16 (m, 7H), 7.10 (d, *J* = 8.9 Hz, 1H), 5.67 (s, 1H), 5.04 (d, *J* = 4.1 Hz, 1H), 4.11–4.06 (m, 1H), 2.23–2.11 (m, 3H), 2.10–1.90 (m, 1H), 1.20 (d, *J* = 3.1 Hz, 3H); ¹³C NMR δ 152.9, 142.2, 131.8, 128.9 (2C), 128.5, 128.3 (2C), 128.2 (2C), 127.2, 126.2, 123.0, 122.8, 118.9, 113.5, 85.1, 80.1, 79.5, 55.7, 51.5, 30.6, 29.4, 2.9; MS *m*/*z* (%) 339 (M⁺, 10.8), 231 (100). Anal. Calcd for C₂₄H₂₁NO: C, 84.92; H, 6.24; N, 4.13. Found: C, 84.75; H, 6.40; N, 4.14.

Preparation of (7a*R*,10*R*,12*S*)-7a,8,9,10-Tetrahydro-10phenylethynyl-12-phenyl-12*H*-naphtho[1,2-e]pyrrolo[2,1-*b*]-[1,3]oxazine (11d). By reaction of 10 with PhC≡CMgCl at 0 °C for 1 h, 11d was obtained in 90% yield. Mp 166−168 °C (Et₂O); $[α]^{25}_{D}$ +488.1 (*c* 0.2, CHCl₃); IR *v* 3467, 2948, 2208, 1623, 1597 cm⁻¹; ¹H NMR δ 7.80−7.59 (m, 2H), 7.43−7.34 (m, 1H), 7.33−7.15 (m, 7H), 7.13−7.05 (m, 1H), 7.04−6.96 (m, 1H), 6.95−6.84 (m, 2H), 6.46 (d, *J* = 7.2 Hz, 2H), 5.74 (s, 1H), 5.12 (d, *J* = 3.8 Hz, 1H), 4.57−4.53 (m, 1H), 2.40−2.18 (m, 3H), 2.16−2.01 (m, 1H); ¹³C NMR δ 153.1, 143.1, 131.8, 131.0 (2C), 129.1, 128.7 (2C), 128.6, 128.4 (2C), 128.3, 127.6 (2C), 127.4, 127.2, 126.4, 123.1, 122.9, 122.6, 118.9, 113.2, 90.0, 85.3, 83.8, 55.5, 52.4, 31.1, 29.7; MS *m/z* (%) 401 (M⁺, 11.7), 231 (100). Anal. Calcd for C₂₉H₂₃NO: C, 86.75; H, 5.77; N, 3.49. Found: C, 86.49; H, 5.68; N, 3.54.

Preparation of (2*S*,5*R*)-2,5-Diallylpyrrolidine-1-carboxylic Acid *tert*-Butyl Ester (3a). To a stirred solution of $H_2C=CHCH_2$ -MgBr made from Mg (1.20 g, 50 mmol) and $H_2C=CHCH_2Br$ (5.05 g, 50 mmol) in anhydrous Et₂O (30 mL) was added dropwise a solution of 10 (2.09 g, 5 mmol) in anhydrous toluene (50 mL) at -78 °C within 10 min. Then a saturated aqueous solution of NH₄Cl (30 mL) was added to quench the reaction at 0 °C. The separated organic layer was washed with H_2O and brine, then dried over anhydrous Na_2SO_4 . Removal of the solvent gave the crude product **12b** as a yellowish solid (1.84 g, 96%), which was used in the next step without characterization.

To an aqueous solution of NaOH (6.0 M, 1 mL), THF (2 mL), and MeOH (2 mL) was added the crude **12b** (384 mg, 1 mmol). The resulting mixture was heated at 60 °C for 1 h and then cooled to room temperature. (Boc)₂O (655 mg, 3 mmol) was then added and the mixture was stirred for another 0.5 h. After the organic solvent was removed, the residue was diluted with Et₂O (20 mL) and H₂O (20 mL). The separated organic layer was washed with brine and dried over anhydrous Na₂SO₄. After the removal of the solvent, the residue was purified by chromatography (silica gel, EtOAc-PE) to give product **3a** (219 mg, 87%). IR *v* 2975, 1693, 1390 cm⁻¹; ¹H NMR (50 °C) δ 5.80–5.71 (m, 2H), 5.07–5.02 (m, 4H), 3.82 (br, 2H), 2.54 (br, 2H), 2.16–2.10 (m, 2H), 1.89–1.83 (m, 2H), 1.70–1.65 (m, 2H), 1.47 (s, 9H); ¹³C NMR (50 °C) δ 154.7, 135.3 (2C), 116.6 (2C), 79.0, 58.2 (2C), 39.8 (2C), 28.5 (5C); MS *m*/*z* (%) 252 (M + 1, 2.18), 57 (100). Anal. Calcd for C₁₅H₂₅NO₂: C, 71.67; H, 10.02; N, 5.57. Found: C, 71.49; H, 10.21; N, 5.50.

Preparation of (2S,5R)-2-Allyl-5-vinylpyrrolidine-1-carboxylic Acid tert-Butyl Ester (3b). To a stirred solution of 11a (655 mg, 2.0 mmol) in anhydrous Et₂O (20 mL) was added dropwise a solution of H₂C=CHCH₂MgBr made from Mg (0.24 g, 10 mmol) and H₂C=CHCH₂Br (1.21 g, 10 mmol) in anhydrous Et₂O (15 mL) at -40 °C. After the reaction was stirred at -40 °C for 1.0 h (monitored by TLC), a saturated aqueous solution of NH₄Cl (10 mL) was added to quench the reaction. The separated organic layer was washed with H₂O and brine, then dried over anhydrous Na₂SO₄. Removal of the solvent gave the crude product 12c as a yellowish solid (709 mg, 96%), which was used in the next step without characterization.

To an aqueous solution of NaOH (6.0 M, 1 mL), THF (2 mL), and MeOH (2 mL) was added the crude 12c (370 mg, 1.0 mmol). The resulting mixture was heated at 60 °C for 1 h, and then cooled to room temperature. (Boc)₂O (655 mg, 3 mmol) was then added and the mixture was stirred for another 0.5 h. After the organic solvent was removed, the residue was diluted with Et₂O (20 mL) and H₂O (20 mL). The separated organic layer was washed with brine and dried over anhydrous Na₂SO₄. After the removal of the solvent, the residue was purified by chromatography (silica gel, EtOAc-PE) to give the desired product 3b (209 mg, 88%) as a yellow oil; $[\alpha]^{25}_{D}$ -45.1 (c 0.2, CHCl₃) {lit.^{6b} $[\alpha]^{20}_{D}$ +40.7 (c 0.2, CHCl₃) for the enantioisomer}; IR v 3077, 2975, 1692, 1641, 1478 cm⁻¹; ¹H NMR (50 °C) δ 5.81–5.72 (m, 2H), 5.15–5.00 (m, 4H), 4.28 (br, 1H), 3.86 (br, 1H), 2.63 (br, 1H), 2.15-2.08 (m, 1H), 2.02-1.94 (m, 1H), 1.93–1.87 (m, 1H), 1.73–1.64 (m, 2H), 1.45 (s, 9H); ¹³C NMR (50 °C) δ 154.7, 140.1, 135.3, 116.6, 113.7, 77.2, 60.5, 58.2, 39.7, 30.4, 28.9, 28.4 (3C); MS m/z (%) 273 (M⁺, 0.15), 96 (100). Anal. Calcd for C₁₄H₂₃NO₂: C, 70.85; H, 9.77; N, 5.90. Found: C, 70.56; H, 9.73; N, 5.98. By a similar procedure as that used for 3b, compounds 3c-f were

prepared.

Preparation of (2*S*,*SR*)-2-Allyl-5-ethynylpyrrolidine-1-carboxylic Acid *tert*-Butyl Ester (3c). From 11b and H₂C=CHCH₂MgBr, 3c was obtained as a yellowish oil, $[\alpha]^{25}_{D} + 32.6$ (*c* 0.2, CHCl₃) {lit.^{6b} $[\alpha]^{20}_{D}$ -30 (*c* 0.2, CH₃Cl₂) for the enantioisomer}; IR *v* 3295, 2978, 2117, 1685, 1639 cm⁻¹; ¹H NMR δ 5.80–5.64 (m, 1H), 5.14–5.05 (m, 2H), 4.51 (br, 1H), 3.83 (br, 1H), 2.63 (br, 1H), 2.35–2.20 (m, 2H), 2.08–1.80 (m, 4H), 1.52 (s, 9H); ¹³C NMR δ 153.7, 134.7, 116.9, 84.5, 79.6, 69.6, 57.5, 48.4, 39.5, 38.8, 31.6, 29.6, 28.8, 28.2 (3C); MS *m*/*z* (%) 194 (23.92), 57 (100). Anal. Calcd for C₁₄H₂₁NO₂: C, 71.46; H, 8.99; N, 5.95. Found: C, 71.29; H, 9.13; N, 6.07.

Preparation of (2*S*,5*R*)-2-(2-Methyl-2-propenyl)-5-ethynylpyrrolidine-1-carboxylic Acid *tert*-Butyl Ester (3d). From 11b and H₂C=C(Me)CH₂MgBr, 3d was obtained as a yellowish oil. $[α]^{25}_{D}$ +14.6 (*c* 0.2, CHCl₃); IR ν 3252, 2976, 2935, 2123, 1781, 1690 cm⁻¹; ¹H NMR δ 4.75 (d, *J* = 11.3 Hz, 2H), 4.49 (br, 1H), 3.96 (br, 1H), 2.68 (br, 1H), 2.26–2.27 (m, 1H), 1.80–2.18 (m, 5H), 1.76 (s, 3H), 1.48 (s, 9H); ¹³C NMR δ 153.6, 142.7, 112.4, 84.6, 79.5, 69.5, 56.3, 48.1, 43.1, 31.6, 29.2, 28.2 (3C), 22.1; MS *m*/*z* (%) 249 (M⁺, 0.02), 57 (100). Anal. Calcd for C₁₅H₂₃NO₂: C, 72.25; H, 9.30; N, 5.62. Found: C, 72.10; H, 9.35; N, 5.74.

Preparation of (2*S*,5*R*)-2-Allyl-5-(prop-1-ynyl)-pyrrolidine-1-carboxylic Acid *tert*-Butyl Ester (3e). From 11c and H₂C=CHCH₂MgBr, 3e was obtained as a yellowish oil. $[\alpha]^{25}_{D}$ +28.9 (*c* 0.2, CHCl₃); IR *ν* 3292, 2978, 2237, 1783, 1713 cm⁻¹; ¹H NMR (50 °C) δ 5.86-5.72 (m, 1H), 5.11-5.01 (m, 2H), 4.48 (br, 1H), 3.84-3.75 (m, 1H), 2.66-2.60 (m, 1H), 2.31-2.20 (m, 1H), 2.02-1.83 (m, 4H), 1.80-1.79 (m, 3H), 1.48 (s, 9H); ¹³C NMR (50 °C) δ 154.0, 135.2, 116.6, 80.1, 79.4, 77.2, 57.6, 49.1, 39.4, 32.2, 29.4, 28.5 (3C), 3.4; MS *m/z* (%) 235 (M, 0.01), 84 (92.99), 57 (100). Anal. Calcd for C₁₅H₂₃NO₂: C, 72.25; H, 9.30; N, 5.62. Found: C, 72.04; H, 9.47; N, 5.48.

Preparation of (2*S*,*SR*)-2-Allyl-5-phenylethynylpyrrolidine-1-carboxylic Acid *tert*-Butyl Ester (3f). From 11d and H₂C=CHCH₂MgBr, 3f was obtained as a yellowish oil. $[α]^{25}{}_{D}$ +37.7 (*c* 0.2, CHCl₃); IR *v* 3075, 2976, 2214, 1783, 1720, 1695 cm⁻¹; ¹H NMR (50 °C) δ 7.32–7.30 (m, 2H), 7.19–7.17 (m, 3H), 5.80–5.71 (m, 1H), 5.04–4.93 (m, 2H), 4.66 (br, 1H), 3.81 (br, 1H), 2.61–2.54 (m, 1H), 2.32–2.27 (m, 1H), 2.06–1.96 (m, 2H), 1.95–1.83 (m, 2H), 1.40 (s, 9H); ¹³C NMR (50 °C) δ 154.0, 135.0, 131.5 (2C), 128.1 (2C), 127.9, 123.5, 116.9, 90.5, 81.9, 79.7, 57.7, 49.5, 39.2, 32.2, 29.5, 28.5 (3C); MS *m/z* (%) 311 (M⁺, 1.08), 170 (100). Anal. Calcd for C₂₀H₂₅NO₂: C, 77.14; H, 8.09; N, 4.50. Found: C, 76.87; H, 8.02; N, 4.68.

Preparation of (1*R*,5*S*)-8-Azabicyclo[3.2.1]-2-octene-8carboxylic Acid *tert*-Butyl Ester (4). To a stirred solution of 3b (2.37 g, 10 mmol) in CH₂Cl₂ (15 mL) was added Grubbs 2nd generation catalyst (212 mg, 0.25 mmol) in one portion at room temperature under N₂. After the resulting mixture was refluxed for 12 h, it was cooled to room temperature and the solvent was evaporated under vacuum. The residue was purified by chromatography (silica gel, 16% EtOAc in PE) to give the product 4 (1.98 g, 95%) as colorless crystals. Mp 35–36 °C (hexane–EtOAc), $[\alpha]^{25}_{D}$ +3.6 (*c* 0.2, CHCl₃) {lit.^{5d} [α]²⁰_D+3.0 (*c* 1.0, CHCl₃)}; IR (KBr) *v* 2977, 1697, 1391 cm⁻¹; ¹H NMR δ 5.98 (br, 1H), 5.50–5.53 (m, 1H), 4.32 (br, 1H), 4.25 (br, 1H), 2.65–2.85 (m, 1H), 2.15 (br, 1H), 1.60–2.05 (m, 4H), 1.45 (s, 9H); ¹³C NMR δ 154.0, 133.0, 132.4, 123.9, 123.4, 79.0, 53.5, 52.7, 51.8, 34.9, 34.5, 34.2, 33.9, 30.2, 29.0, 28.3 (3C); MS *m*/*z* (%) 210 (M + 1, 1.03), 57 (100).

Preparation of (3aS,3bR,5aS,6aS)-3-Bromo-7-azabicyclo-[3.2.1]octane[2,3-*d*]isoxazole-7-carboxylic Acid *tert*-Butyl Ester (5b). The mixture of 4 (2.1 g, 10 mmol), Br₂C=NOH (4.2 g, 20 mmol), and NaHCO₃ (8.4 g, 100 mmol) in EtOAc (30 mL) was stirred vigorously at 10 °C for 40 h. After the reaction was finished (monitored by TLC), H₂O (50 mL) was added. The organic layer was separated and dried over anhydrous Na₂SO₄. After the removal of the solvent, the residue was purified by chromatography (silica gel, 20% EtOAc in PE) to give **5b** (3.0 g, 81%) as colorless crystals. Mp 103–104 °C (hexane−EtOAc), $[α]^{25}_{D}$ +3.7 (*c* 0.2, CHCl₃); IR ν 2973, 2930, 1697, 1378 cm⁻¹; ¹H NMR δ 4.93−5.01 (m, 1H), 4.69 (br, 1H), 4.38 (br, 1H), 3.35 (d, 1H, *J* = 8.94 Hz), 1.55−2.19 (m, 6H), 1.48 (s, 9H); ¹³C NMR δ 152.7, 141.0, 80.1, 75.8, 58.0, 51.8, 50.3, 36.6, 29.8, 28.2 (3C), 27.2; MS *m*/*z* (%) 330 (M⁺, 0.04), 57 (100). Anal. Calcd for C₁₃H₁₉BrN₂O₃: C, 47.14; H, 5.78; N, 8.46. Found: C, 47.31; H, 5.69; N, 8.52.

Preparation of (3aS,3bR,5aS,6aS)-3-Methoxyl-7-azabicyclo[3.2.1]octane[2,3-d]isoxazole-7-carboxylic Acid tert-Butyl Ester (13). To a solution of 5a (1.0 g, 3 mmol) in dry MeOH (20 mL) was added NaOMe (810 mg, 15 mmol) at room temperature and the resulting mixture was refluxed for 8 h (monitored by TLC). After the reaction was cooled to 0 °C, H₂O (60 mL) was added. The aqueous mixture was extracted with Et₂O (3 × 20 mL) and the combined organic layers were washed with brine and dried over anhydrous Na₂SO₄. After the removal of the solvent under vacuum, the product 13 (850 mg, 99%) was obtained as a colorless oil, which was pure enough for most analytic purposes. The analytical sample was purified by chromatography (silica gel, 20% EtOAc in PE). $[\alpha]^{25}_{D}$ +36.3 (*c* 0.2, CHCl₃); IR *v* 2975, 1695, 1618, 1415 cm⁻¹; ¹H NMR δ 4.88–4.97 (m, 1H), 4.60 (br, 1H), 4.23–4.39 (m, 1H), 3.87 (s, 3H), 3.31 (d, 1H, *J* = 8.94 Hz), 1.85–2.11 (m, 4H), 1.55–1.78 (m, 2H), 1.46 (s, 9H); ¹³C NMR δ 167.0, 151.9, 78.8, 75.3, 56.6, 51.2, 50.7, 50.0, 36.1, 29.6, 27.7 (3C), 26.4; MS *m/z* (%) 282 (M⁺, 3), 83 (100). Anal. Calcd for C₁₄H₂₂N₂O₄: C, 59.56; H, 7.85; N, 9.92. Found: C, 59.44; H, 7.91; N, 9.83.

Preparation of (1R,2R,3S,5S)-3-Hydroxy-8-azabicyclo[3.2.1]octane-2,8-dicarboxylic Acid 8-tert-Butyl Ester 2-Methyl Ester (14). To a solution of 13 (520 mg, 1.84 mmol) in H₂O-MeOH (1:5 by v/v, 12 mL) was added H₃BO₃ (250 mg, 4 mmol) and wet W-2 Raney nickel (50 mg). The resulting suspension was hydrogenated for 3 h under atmospheric pressure (balloon) and room temperature. After the catalyst was filtered off, the filtrate was diluted with $H_2O(30 \text{ mL})$ and the aqueous mixture was extracted with Et₂O (3×20 mL). The combined organic layers were washed with brine and dried over anhydrous Na2SO4. Removal of the solvent under vacuum gave 14 (530 mg, 100%) as a colorless oil, which was pure enough for most analytic purpose. The analytical sample was purified by chromatography (silica gel, 20% EtOAc in PE). $[\alpha]_{D}^{25}$ –10.9 (c 0.2, CHCl₃); IR v 2990, 1737, 1672, 1419 cm⁻¹; 1 H NMR δ 4.66 (br, 1H), 4.20–4.45 (m, 1H), 3.95–4.10 (m, 1H), 3.70 (s, 3H), 3.17 (d, 1H, J = 10.98 Hz), 2.85–2.91 (m, 1H), 1.80–2.18 (m, 4H), 1.52-1.70 (m, 2H), 1.43 (s, 4.5H), 1.41 (s, 4.5H); ¹³C NMR δ 172.5, 152.3, 79.1, 63.6, 54.6, 52.8, 51.5, 50.9, 37.8, 28.0 (3C), 27.6, 27.3; MS m/z (%) 285 (M⁺, 4), 83 (100). Anal. Calcd for C₁₄H₂₃NO₅: C, 58.93; H, 8.12; N, 4.91. Found: C, 59.01; H, 8.00; N, 4.99.

Preparation of (1R,2R,3S,5S)-3-Benzoyloxy-8-aza-bicyclo-[3.2.1]octane-2,8-dicarboxylic Acid 8-tert-Butyl Ester 2-Methyl Ester (15). To a solution of 14 (390 mg, 1.4 mmol) in DCM (5 mL) was added PhCOCl (300 mg, 2.1 mmol), TEA (710 mg, 7.0 mmol), and 4-dimethylaminopyridine (30 mg, 0.25 mmol). After the mixture was stirred at room temperature for 12 h, it was diluted with DCM (30 mL) and washed with saturated aqueous NaHCO3 (10 mL) and brine (10 mL). The organic phase was dried with anhydrous Na2SO4 and concentrated to afford 15 (510 mg, 96%) as colorless crystals. Mp 128-129 °C (hexane-EtOAc); $[\alpha]^{25}_{D} - 11.4$ (c 0.2, CHCl₃) {lit.^{5b} $[\alpha]^{23}_{D} + 4.3$ (no concentration and solvent were reported) for the enantioisomer}; IR ν 2977, 1744, 1719, 1680, 1410, 1275 cm⁻¹; ¹H NMR δ 7.96–7.98 (m, 2H), 7.53 (t, 1H, *J* = 7.23 Hz), 7.40 (t, 2H, *J* = 7.56 Hz), 5.48 (m, 1H), 4.68 (d, 0.5H, *J* = 5.16 Hz), 4.55 (s, 1H), 4.37 (s, 0.5H), 3.70 (s, 1.5H), 3.69 (s, 1.5H), 3.07 (d, 1H, I = 5.49 Hz, 2.57 (td, 1H, I = 12.0, 3.0 Hz), 1.77–2.14 (m, 5H), 1.47 (s, 4.5H), 1.45 (s, 4.5H); ¹³C NMR δ 170.1, 170.0, 165.7, 152.4, 152.0, 133.0, 129.8, 129.5, 128.3, 79.5, 79.4, 66.6, 54.4, 52.6, 51.7, 51.5, 49.0, 48.8, 33.4, 33.3, 28.8, 28.3 (3C), 28.0, 27.8, 27.1; MS m/z (%) 389 (M⁺, 2), 57 (100). Anal. Calcd for C21H27NO6: C, 64.77; H, 6.99; N, 3.60. Found: C, 64.63; H, 7.03; N. 3.81.

Preparation of (–)-**Cocaine (1).** To a solution of **15** (510 mg, 1.3 mmol) in CH₂Cl₂ (15 mL) was added dropwise CF₃CO₂H (5.0 g, 43.9 mmol) and the resulting solution was stirred at room temperature for 1 h. After the reaction system was diluted by CH₂Cl₂ (15 mL), saturated aqueous NaHCO₃ was added dropwise. The organic layer was separated and the aqueous layer was extracted with CH₂Cl₂ (4 × 10 mL). The combined organic layers were washed with brine and dried over anhydrous Na₂SO₄. After removal of the solvent under vacuum, the crude amine was obtained as a yellow oil.

To a solution of the crude amine and formaldehyde (37% aqueous solution, 0.6 mL, 8 mmol) in MeCN (20 mL) was added NaBH₃CN (170 mg, 2.7 mmol). Then the resulting mixture was stirred for 1 h at room temperature and acidified to pH 6 with HOAc. After the mixture was stirred for an additional 0.5 h, it was neutralized to pH 9 with ammonia. The mixture was diluted by addition of CH_2Cl_2 (30 mL) and saturated aqueous NaHCO₃ (30 mL). The organic layer was separated

and the aqueous layer was extracted with CH₂Cl₂ (5 × 10 mL). The combined organic layers were washed with brine and dried over anhydrous Na₂SO₄. After removal of the solvent under vacuum, the residue was purified by chromatography (silica gel, Et₃N:MeOH: CH₂Cl₂ = 0.25:1:100) to give (–)-cocaine (1) (330 mg, 85%) as color-less crystals. Mp 97–98 °C (Et₂O) (lit.^{5d} mp 93–94 °C); $[\alpha]^{25}_{D}$ –16.5 (*c* 0.2, CHCl₃) {lit.^{5d} $[\alpha]^{25}_{D}$ –16.2 (*c* 1.2, CHCl₃)}; IR *v* 2946, 1737, 1709 cm⁻¹; ¹H NMR δ 8.02 (dd, 2H, *J* = 8.22, 1.38 Hz), 7.53 (tt, 1H, *J* = 7.56, 1.51 Hz), 7.42 (t, 2H, *J* = 7.56 Hz), 5.25 (m, 1H), 3.71 (s, 3H), 3.56–3.57 (m, 1H), 3.30 (br, 1H), 3.00–3.03 (m, 1H), 2.43 (td, 1H, *J* = 10.35, 2.70 Hz), 2.23 (s, 3H), 2.08–2.21 (m, 2H), 1.85–1.90 (m, 1H), 1.70–1.73 (m, 2H); ¹³C NMR δ 170.7, 166.1, 132.8, 130.2, 129.6, 128.2, 66.8, 64.8, 61.5, 51.3, 50.1, 41.1, 35.5, 25.4, 25.2; MS *m/z* (%) 303 (M⁺, 4), 82 (100).

Preparation of (1R,2R,3S,5S)-2-Cyano-3-hydroxy-8-azabicyclo[3.2.1]octane-8-carboxylic Acid tert-Butyl Ester (16). To a solution of **5b** (330 mg, 1.0 mmol) in H_2O -MeOH (1:5 by v/v, 10 mL) was added H₃BO₃ (190 mg, 3 mmol) and wet W-2 Raney nickel (40 mg). The resulting suspension was hydrogenated for 3 h under atmospheric pressure (balloon) and room temperature. After the catalyst was filtered off, the filtrate was diluted with H2O (30 mL) and the mixture was extracted with Et₂O (3 \times 20 mL). The combined organic layers were washed with brine and dried over anhydrous Na2SO4. Removal of the solvent under vacuum gave 16 (250 mg, 99%) as colorless crystals. Mp 140–141 °C (hexane–EtOAc) (lit.^{5d} mp 139–140 °C); $[\alpha]_{D}^{25}$ –10.2 (c 0.2, CHCl₃) {lit.^{5d} $[\alpha]_{D}^{25}$ –9.0 (c 1.0, CHCl₃)}; IR *ν* 2977, 2240, 1685 cm⁻¹; ¹H NMR δ 4.53–4.61 (m, 1H), 4.32–4.45 (m, 1H), 4.15–4.26 (m, 1H), 3.71 (d, 1H, J = 5.16 Hz), 3.12 (br, 1H), 1.75–2.05 (m, 4H), 1.58–1.65 (d, 2H, J = 7.56 Hz), 1.49 (s, 9H); ¹³C NMR δ 152.2, 117.8, 80.4, 62.6, 53.7, 51.5 41.5, 37.6, 28.1 (3C), 27.5 (2C); MS m/z (%) 252 (M⁺, 0.65), 57 (100).

Preparation of (1R,5S)-2-Cyano-8-azabicyclo[3.2.1]-2-octene-8-carboxylic Acid tert-Butyl Ester (17). To a solution of 16 (1.30 g, 5.20 mmol) in CH₂Cl₂ (40 mL) was successively added Et₃N (1.04 g, 10.4 mmol) and MeSO₂Cl (1.2 g, 1.04 mmol) at 0 °C. After the resulting mixture was stirred at 0 °C for 1 h, it was allowed to stir at room temperature for another 12 h. Then it was quenched with ice-water and extracted with CH_2Cl_2 (3 × 20 mL). The combined organic layers were washed with H₂O and brine and dried over Na₂SO₄. After removal of the solvent, the residue was purified by chromatography (silica gel, 16% EtOAc) to give pure 17 (1.04 g, 85%) as colorless crystals. Mp 96–97 °C (hexane–EtOAc); $[\alpha]^{25}_{D}$ –94.4 (c 0.2, CHCl₃); IR ν 2971, 2209, 1693 cm $^{-1};~^{1}\mathrm{H}$ NMR δ 6.50 (br, 1H), 4.45 (br, 1H), 4.34 (br, 1H), 2.87 (br, 1H), 1.97-2.35 (m, 4H), 1.62-1.75 (m, 1H), 1.46 (s, 9H); ¹³C NMR δ 153.2, 141.9, 118.8, 116.8, 80.0, 54.8, 50.9, 34.4, 33.6, 29.1, 27.9 (3C); MS m/z (%) 234 (M⁺, 0.10), 57 (100). Anal. Calcd for C13H18N2O2: C, 66.64; H, 7.74; N, 11.96. Found: C, 66.77; H, 7.80; N, 12.10.

Preparation of (1R,5S)-2-Acetyl-8-azabicyclo[3.2.1]-2-octene-8-carboxylic Acid tert-Butyl Ester (18). To a stirred solution of 17 (230 mg, 1.0 mmol) in Et_2O (10 mL) was added a solution of MeLi in Et₂O (1.6 M, 3.0 mL, 4.80 mmol) dropwise at 0 °C under N₂. After the resulting mixture was warmed to room temperature and stirred for 2 h, it was quenched with H₂O (3 mL) at 0 °C. PPTS (10 mg, 0.04 mmol) was then added and the new mixture was stirred for another 12 h at room temperature. The organic layer was separated and the aqueous layer was extracted with Et₂O (3 \times 30 mL). The combined organic layers were washed with brine and dried over Na2SO4. After removal of the solvent, the residue was purified by chromatography (silica gel, 16% EtOAc) to give pure 18 (220 mg, 87%) as colorless crystals. Mp 62-63 °C (hexane-EtOAc) (lit.^{6f} mp 64-65 °C); $[\alpha]_{D}^{25}$ -114.5 (c 0.2, CHCl₃) {lit.^{6f} $[\alpha]_{D}^{25}$ -126.8 (c 1.0, CHCl₃)}; IR v 2977, 1699, 1660 cm⁻¹; ¹H NMR δ (as rotamers) 6.67 (s, 1H), 4.91 (d, 1H, J = 5.83 Hz), 4.35 (s, 1H), 2.93 (m, 1H), 2.27 (s, 3H), 2.10–2.20 (m, 3H), 1.80 (dt, 1H, J = 2.4, 10.98 Hz), 1.50–1.61 (m, 1H), 1.43 (s, 9H); ¹³C NMR δ (as rotamers) 196.2, 153.6 (0), 145.3, 137.2, 79.4, 51.9, 51.1, 34.7, 34.5, 29.5, 28.2 (3C), 24.7; MS *m*/*z* (%) 251 (M⁺, 0.50), 57 (100).

Preparation of (–)-Ferruginine (2). To a solution of 18 (330 mg, 1.3 mmol) in CH_2Cl_2 (15 mL) was added dropwise CF_3CO_2H (5.0 g, 43.86 mmol) and the resulting mixture was stirred for 1 h at room temperature. After the reaction was diluted by CH_2Cl_2 (15 mL), saturated aqueous NaHCO₃ was added dropwise. The organic layer was separated and the aqueous layer was extracted with CH_2Cl_2 (4 × 10 mL). The combined organic layers were washed with brine and dried over anhydrous Na₂SO₄. After removal of the solvent under vacuum, the crude amine product was obtained as a yellow oil.

To a solution of the crude amine product and formaldehyde (37% aqueous solution, 0.6 mL, 8 mmol) in MeCN (20 mL) was added NaBH₃CN (170 mg, 2.7 mmol). Then the resulting mixture was stirred for 1 h at room temperature and acidified to pH 6 with HOAc. After the mixture was stirred for an additional 0.5 h, it was neutralized to pH 9 with ammonia. The mixture was diluted by addition of CH₂Cl₂ (30 mL) and saturated aqueous NaHCO3 (30 mL). The organic layer was separated and the aqueous layer was extracted with CH_2Cl_2 (5 × 10 mL). The combined organic layers were washed with brine and dried over anhydrous Na₂SO₄. After removal of the solvent under vacuum, the residue was purified by chromatography (silica gel, Et₃N:MeOH: $CH_2Cl_2 = 0.5:10:100$) to give (-)-ferruginine (2) (200 mg, 91%) as a yellow oil. $[\alpha]_{D}^{25}$ – 52.5 (c 0.2, CHCl₃) {lit.^{6e} $[\alpha]_{D}^{25}$ – 50.8 (c 0.94, CHCl₃); IR ν 2942, 1662 cm⁻¹; ¹H NMR δ 6.73 (br, 1H), 3.90 (d, 1H, *J* = 4.47), 3.23 (br, 1H), 2.72 (d, 1H, *J* = 20.79 Hz), 2.32 (s, 3H), 2.26 (s, 3H), 2.05–2.20 (m, 2H), 1.93 (dd, 1H, J = 19.59, 3.78 Hz), 1.70 (t, 1H, J = 9.63 Hz), 1.48 (t, 1H, J = 8.91 Hz); ¹³C NMR δ (as rotamers) 196.9, 143.1, 136.3, 57.0, 56.9, 36.7, 33.2, 32.4, 29.1, 24.4; MS m/z (%) 165 (M⁺, 22), 57 (100).

ASSOCIATED CONTENT

Supporting Information. ¹H and ¹³C NMR spectra for **11a**-**d**, **3a**-**f**, **4**, **5b**, **13**-**18**, **1**, and **2**, as well as the crystallographic data of compound **5b**. This material is available free of charge via the Internet at http://pubs.acs.org.

AUTHOR INFORMATION

Corresponding Author

*E-mail: wangxinyan@mail.tsinghua.edu.cn and yfh@mail.tsinghua. edu.cn. Phone: +86-10-62795380. Fax: +86-10-62771149.

ACKNOWLEDGMENT

This work was supported by NNSFC (30600779, 20672066) and by CFKSTIP from Education Ministry of China (706003).

REFERENCES

 (a) Carroll, F. I.; Lewin, A. H.; Boja, J. W.; Kuharf, M. J. J. Med. Chem. 1992, 35, 969–981.
 (b) Koob, G. F.; Bloom, F. E. Science 1988, 242, 715–723.

(2) (a) Daly, D. W. *Cell. Mol. Neurobiol.* **2005**, *25*, 513–552. (b) Gohlke, H.; Gündish, D.; Schwartz, S.; Seitz, G.; Tilotta, M. C.; Wegge, T. J. Med. Chem. **2002**, *45*, 1064–1072.

(3) (a) Pollini, G. P.; Benetti, S.; De Risi, C.; Zanirato, V. Chem. Rev. 2006, 106, 2434–2454. (b) Singh, S. Chem. Rev. 2000, 100, 925–1024.

(4) Campbell, H. F.; Edwards, O. E.; Kolt, R. Can. J. Chem. 1977, 55, 1372-1379.

(5) (a) Davis, F. A.; Theddu, N.; Edupuganti, R. Org. Lett. 2010,
12, 4118–4121. (b) Mans, D. M.; Pearson, W. H. Org. Lett. 2004,
6, 3305–3308. (c) Lee, J. C.; Lee, K.; Cha, J. K. J. Org. Chem. 2000,

65, 4773-4775. (d) Lin, R. H.; Castells, J.; Rapoport, H. J. Org. Chem. 1998, 63, 4069-4078.

(6) (a) Piccardi, R.; Renaud, P. Eur. J. Org. Chem. 2007, 4752–4757.
(b) Aggarwal, V. K.; Astle, C. J.; Rogers-Evans, M. Org. Lett. 2004, 6, 1469–1471. (c) Katoh, T.; Kakiya, K.; Nakai, T.; Nakamura, S.; Nishide, K.; Node, M. Tetrahedron: Asymmetry 2002, 13, 2351–2358.
(d) Gauthier, I.; Royer, J.; Husson, H.-P. J. Org. Chem. 1997, 62, 6704–6705. (e) Davies, H. M. L.; Matasi, J. J.; Hodges, L. M.; Huby, N. J. S.; Thornley, C.; Kong, N.; Houser, J. H. J. Org. Chem. 1997, 62, 1095–1105. (f) Hernandez, A. S.; Thaler, A.; Castells, J.; Rapoport, H. J. Org. Chem. 1996, 61, 314–323. (g) Rigby, J. H.; Pigge, F. C. J. Org. Chem. 1995, 60, 7392–7393.

(7) (a) Kaliappan, K. P.; Das, P.; Chavan, S. T.; Sabharwal, S. G.
 J. Org. Chem. 2009, 74, 6266–6274. (b) Kuznetsov, N. Y.; Khrustalev,
 V. N.; Godovikov, I. A.; Bubnov, Y. N. Eur. J. Org. Chem. 2006, 113–120.

(8) (a) Brenneman, J. B.; Martin, S. F. Org. Lett. 2004, 6, 1329–1331.
(b) Brenneman, J. B.; Machauer, R.; Martin, S. F. Tetrahedron 2004, 60, 7301–7314.

(9) (a) Shu, H.; Noble, A. R.; Zhang, S.; Miao, L.; Trudell, M. L. *Tetrahedron* **2010**, *66*, 4428–4433. (b) Hong, Z.; Liu, L.; Sugiyama, M.; Fu, Y.; Wong, C.-H. *J. Am. Chem. Soc.* **2009**, *131*, 8352–8353. (c) Zhang, S.; Xu, L.; Miao, L.; Shu, H.; Trudell, M. L. *J. Org. Chem.* **2007**, *72*, 3133–3136. (d) Weihofen, R.; Dahnz, A.; Tverskoy, O.; Helmchen, G. *Chem. Commun.* **2005**, 3541–3543. (e) Weeresakare, G. M.; Liu, Z.; Rainier, J. D. *Org. Lett.* **2004**, *6*, 1625–1627.

(10) (a) Cheng, G.; Wang, X.; Su, D.; Liu, H.; Liu, F.; Hu, Y. J. Org.
 Chem. 2010, 75, 1911–1916. (b) Liu, H.; Su, D.; Cheng, G.; Xu, J.;
 Wang, X.; Hu, Y. Org. Biomol. Chem. 2010, 1899–1904.

(11) For the preparation and application of intermediate 6, also see:
(a) Su, D.; Wang, X.; Shao, C.; Xu, J.; Hu, Y. J. Org. Chem. 2011, 76, 188–194.
(b) Dong, Y.; Li, R.; Lu, J.; Xu, X.; Wang, X.; Hu, Y. J. Org. Chem. 2005, 70, 8617–8620.
(c) Wang, X.; Dong, Y.; Sun, J.; Li, R.; Xu, X.; Hu, Y. J. Org. Chem. 2005, 70, 1897–1900.
(d) Xu, X.; Lu, J.; Li, R.; Ge, Z.; Dong, Y.; Hu, Y. Synlett 2004, 122–124.

(12) For selected recent references, see: (a) Dallanoce, C.; Frigerio, F.; Martelli, G.; Grazioso, G.; Matera, C.; Pome, D. Y.; Pucci, L.; Clementi, F.; Gotti, C.; De Amici, M. *Bioorg. Med. Chem.* **2010**, *18*, 4498–4508. (b) Girardin, M.; Alsabeh, P. G.; Lauzon, S.; Dolman, S. J.; Ouellet, S. G.; Hughes, G. Org. Lett. **2009**, *11*, 1159–1162. (c) Pinto, A.; Conti, P.; De Amici, M.; Tamborini, L.; Madsen, U.; Nielsen, B.; Christesen, T.; Brauner-Osborne, H.; De Michelia, C. J. Med. Chem. **2008**, *51*, 2311–2315. (d) Conti, P.; De Amici, M.; Roda, G.; Pinto, A.; Tamborini, L.; Madsen, U.; Nielsen, B.; Brauner-Osborne, H.; De Michelia, C. *Tetrahedron* **2007**, *63*, 2249–2256. (e) Conti, P.; Caligiuri, A.; Pinto, A.; Roda, G.; Tamborini, L.; Nielsen, B.; Madsen, U. Eur. J. Med. Chem. **2007**, *42*, 1059–1068. (f) Conti, P.; De Amici, M.; Pinto, A.; Tamborini, L.; Grazioso, G.; Frolund, B.; Nielsen, B.; Thomsen, C.; Ebert, B.; De Micheli, C. Eur. J. Org. Chem. **2006**, 5533–5542. (g) Proemmel, S.; Wartchow, R.; Hoffmann, H. M. R. *Tetrahedron* **2002**, *58*, 6199–6206.

(13) For selected recent references, see: (a) Caetano, V. F.; Demnitz, F. W. J.; Diniz, F. B.; Mariz, R. M., Jr.; Navarro, M. *Tetrahedron Lett.*2003, 44, 8217–8220. (b) Trost, B. M.; Li, L.; Guile, S. D. J. Am. Chem.
Soc. 1992, 114, 8745–8747. (c) Takahide, N.; Yasuhiro, M. *Heterocycles*1989, 29, 1835–42. (d) Curran, D. P.; Scanga, S. A.; Fenk, C. J. J. Org. Chem. 1984, 49, 3474–3478.

(14) For selected references for β-hydroxy nitriles, see: (a) Maimone, T. J.; Shi, J.; Ashida, S.; Baran, P. S. J. Am. Chem. Soc. **2009**, 131, 17066–17067. (b) Kociolek, M. G.; Kalbarczyk, K. P. Synth. Commun. **2004**, 34, 4387–4394. (c) Bacher, E.; Demnitz, F. W. J.; Hurni, T. Tetrahedron **1997**, 53, 14317–14326. (d) Seo, M. H.; Lee, Y. Y.; Goo, Y. M. Synth. Commun. **1994**, 24, 1433–1439. (e) Wade, P. A.; Bereznak, J. F.; Palfey, B. A.; Carroll, P. J.; Dailey, W. P.; Sivasubramanian, S. J. Org. Chem. **1990**, 55, 3045–3051.

(15) Curran, D. P. J. Am. Chem. Soc. 1983, 105, 5826-5833.

(16) Fleming, F. F.; Shook, B. C. J. Org. Chem. 2002, 67, 3668-3672.